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convert the moles of substance a to the moles of substance b
convert it to the moles of sulfur trioxide
react completely with four point seven moles of sulfur dioxide
put the two moles of so₂ on the bottom
given the moles of propane
convert it to the grams of substance
convert from moles of co₂ to grams
react completely with five moles of o₂
convert the grams of propane to the moles of propane
use the molar ratio

start with 38 grams of H_2O
converted in moles of water to moles of CO_2
using the molar mass of substance b
convert that to the grams of aluminum chloride
add the atomic mass of one aluminum atom
change it to the moles of aluminum
change it to the grams of chlorine
find the molar mass
perform grams to gram conversion

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