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Acid-Catalyzed Dehydration of an Alcohol

Dehydration of Secondary and Tertiary Alcohols are E1 Reactions

Dehydration is a Regioselective Reaction

The More Stable Alkene Has the More Stable Transition State Leading to Its Formation

Dehydration is a Reversible Reaction

Tertiary Alcohols are the Easiest to Dehydrate

An E1 Reaction Creates a Carbocation Intermediate, so Watch Out for Carbocation Rearrangements

Dehydration of a Primary Alcohol is an E2 Reaction

The Major Product is 2-Butene

Dehydration is Stereoselective

Dehydration Under Mild Conditions

The Mechanism

Another Way to Do an E2 Dehydration

Oxidation of Secondary Alcohols

Oxidation of Primary Alcohols

Tertiary Alcohols Cannot Be Oxidized to a Carbonyl Compound

Oxidation by Hypochlorous Acid (HOCl)

The Swern Oxidation

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Conclusion

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intro

my study methods

chemistry

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