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Oxidation and Reduction Electrolytic Cells Non-Standard-State Cell Potentials Atomic Theory Atomic Weight Types of Bonds Periodic Trends Electronegativity Quantum Chemistry Basic Electron Charges Components of Atomic Structure The Wave Mechanical Model Subshells and Electron Configuration Double and Triple Bonds Organic Chemistry: Nomenclature and Structure Alkanes Alkenes Dienes Alkynes Alkyl Halides Cyclic Hydrocarbons Aromatic Hydrocarbons Aryl Halides Ethers and Epoxides Alcohols and Glycols Carboxylic Acids Carboxylic Acid Derivatives Esters Amides Arenes Aldehydes and Ketones Amines Phenols and Quinones Structural Isomerism SIX PRACTICE EXAMS "Practice Test 1 " Answer Key Detailed Explanations of Answers "Practice Test 2 " Answer Key Detailed Explanations of Answers "Practice Test 3" Answer Key Detailed Explanations of Answers "Practice Test 4 " Answer Key Detailed Explanations of Answers "Practice Test 5" Answer Key Detailed Explanations of Answers "Practice Test 6 " Answer Key Detailed Explanations of Answers THE PERIODIC TABLE EXCERPT About Research & Education Association Research & Education Association (REA) is an organization of educators, scientists, and engineers specializing in various academic fields. 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CHAPTER 1 - ABOUT THE SAT II: CHEMISTRY SUBJECT TEST ABOUT THIS BOOK This book provides you with an accurate and complete representation of the SAT II: Chemistry Subject Test. Inside you will find a complete course review designed to provide you with the information and strategies needed to do well on the exam, as well as six practice tests based on the actual exam. The practice tests contain every type of question that you can expect to appear on the SAT II: Chemistry test. Following each test you will find an answer key with detailed explanations designed to help you master the test material. ABOUT THE TEST Who Takes the Test and What Is It Used For? Students planning to attend college take the SAT II: Chemistry Subject Test for one of two reasons: (1) Because it is an admission requirement of the college or university to which they are applying; "OR" (2) To demonstrate proficiency in Chemistry. The SAT II: Chemistry exam is designed for students who have taken one year of college preparatory chemistry. Who Administers The Test? The SAT II: Chemistry Subject Test is developed by the College Board and administered by Educational Testing Service (ETS). The test development process involves the assistance of educators throughout the country, and is designed and implemented to ensure that the content and difficulty level of the test are appropriate. When Should the SAT II: Chemistry be Taken? If you are applying to a college that requires Subject Test scores as part of the admissions process, you should take the SAT II: Chemistry Subject Test toward the end of your junior year or at the beginning of your senior year. If your scores are being used only for placement purposes, you may be able to take the test in the spring of your senior year. For more information, be sure to contact the colleges to which you are applying. When and Where is the Test Given? The SAT II: Chemistry Subject Test is administered five times a year at many locations throughout the country; mostly high schools. To receive information on upcoming administrations of the exam, consult the publication Taking the SAT II: Subject Tests, which may be obtained from your guidance counselor or by contacting: College Board SAT Program P.O. Box 6200 Princeton, NJ 08541-6200 Phone: (609) 771-7600 Website: <http://www.collegeboard.com> Is There a Registration Fee? Yes. There is a registration fee to take the SAT II: Chemistry. Consult the publication Taking the SAT II: Subject Tests for information on the fee structure. Financial assistance may be granted in certain situations. To find out if you qualify and to register for assistance, contact your academic advisor. HOW TO USE THIS BOOK What Do I Study First? Remember that the SAT

II: Chemistry Subject Test is designed to test knowledge that has been acquired throughout your education. Therefore, the best way to prepare for the exam is to refresh yourself by thoroughly studying our review material and taking the sample tests provided in this book. They will familiarize you with the types of questions, directions, and format of the SAT II: Chemistry Subject Test. To begin your studies, read over the review and the suggestions for test-taking, take one of the practice tests to determine your area(s) of weakness, and then restudy the review material, focusing on your specific problem areas. The course review includes the information you need to know when taking the exam. Be sure to take the remaining practice tests to further test yourself and become familiar with the format of the SAT II: Chemistry Subject Test.

When Should I Start Studying? It is never too early to start studying for the SAT II: Chemistry test. The earlier you begin, the more time you will have to sharpen your skills. Do not procrastinate! Cramming is not an effective way to study, since it does not allow you the time needed to learn the test material. The sooner you learn the format of the exam, the more comfortable you will be when you take the exam.

FORMAT OF THE SAT II: CHEMISTRY The SAT II: Chemistry is a one-hour exam consisting of 85 multiple-choice questions. The first part of the exam consists of classification questions. This question type presents a list of statements or questions that you must match up with a group of choices lettered (A) through (E). Each choice may be used once, more than once, or not at all. The exam then shifts to relationship analysis questions which you will answer in a specially numbered section of your answer sheet. You will have to determine if each of two statements is true or false and if the second statement is a correct explanation of the first. The last section is composed strictly of multiple-choice questions with choices lettered (A) through (E).

Material Tested The following chart summarizes the distribution of topics covered on the SAT II: Chemistry Subject Test.

Topic	Percentage	Number of Questions
Atomic & Molecular Structure	25%	21 questions
States of Matter	15%	13 questions
Reaction Types	14%	12 questions
Stoichiometry	12%	10 questions
Equilibrium & Reaction Times	7%	6 questions
Thermodynamics	6%	5 questions
Descriptive Chemistry	13%	11 questions
Laboratory	8%	7 questions

The questions on the SAT II: Chemistry are also grouped into three larger categories according to how they test your understanding of the subject material.

Category / Definition / Approximate Percentage of Test

- 1) Factual Recall / Demonstrating a knowledge and understanding of important concepts and specific information / 20%
- 2) Application / Taking a specific principle and applying it to a practical situation / 45%
- 3) Integration / Inferring information and drawing conclusions from particular relationships / 35%

STUDYING FOR THE SAT II: CHEMISTRY It is very important to choose the time and place for studying that works best for you. Some students may set aside a certain number of hours every morning to study, while others may choose to study at night before going to sleep. Other students may study during the day, while waiting on line, or even while eating lunch. Only you can determine when and where your study time will be most effective. Be consistent and use your time wisely. Work out a study routine and stick to it! When you take the practice tests, try to make your testing conditions as much like the actual test as possible. Turn your television and radio off, and sit down at a quiet desk or table free from distraction. Make sure to clock yourself with a timer. As you complete each practice test, score it and thoroughly review the explanations to the questions you answered incorrectly; however, do not review too much at any one time. Concentrate on one problem area at a time by reviewing the questions and explanations, and by studying our review until you are confident you completely understand the material. Keep track of your scores. By doing so, you will be able to gauge your progress and discover general weaknesses in particular sections. You should carefully study the reviews that cover your areas of difficulty, as this will build your skills in those areas.

TEST TAKING TIPS Although you may be unfamiliar with standardized tests such as the SAT II: Chemistry Subject Test, there are many ways to acquaint yourself with this type of examination and help alleviate your test-taking anxieties. Become comfortable with the format of the exam. When you are practicing to take the SAT II: Chemistry Subject Test, simulate the conditions under which you will be taking the actual test. Stay calm and pace yourself. After simulating the test only a couple of times, you will boost your chances of doing well, and you will be able to sit down for the actual exam with much more confidence. Know the directions and format for each section of the test. Familiarizing yourself with the directions and format of the exam will not only save you time, but will also ensure that you are familiar enough with the SAT II: Chemistry Subject Test to avoid nervousness (and the mistakes caused by being nervous). Do your scratchwork in the margins of the test booklet. You will not be given scrap paper during the exam, and you may not perform scratchwork on your answer sheet. Space is provided in your test booklet to do any necessary work or draw diagrams. If you are unsure of an answer, guess. However, if you do guess - guess wisely. Use the process of elimination by going through each answer to a question and ruling out as many of the answer choices as possible. By eliminating three answer choices, you give yourself a fifty-fifty chance of answering correctly since there will only be two choices left from which to make your guess. Mark your answers in the appropriate spaces on the answer sheet.

Fill in the oval that corresponds to your answer darkly, completely, and neatly. You can change your answer, but remember to completely erase your old answer. Any stray lines or unnecessary marks may cause the machine to score your answer incorrectly. When you have finished working on a section, you may want to go back and check to make sure your answers correspond to the correct questions. Marking one answer in the wrong space will throw off the rest of your test, whether it is graded by machine or by hand. You don't have to answer every question. You are not penalized if you do not answer every question. The only penalty results from answering a question incorrectly. Try to use the guessing strategy, but if you are truly stumped by a question, remember that you do not have to answer it. Work quickly and steadily. You have a limited amount of time to work on each section, so you need to work quickly and steadily. Avoid focusing on one problem for too long. Before the Test Make sure you know where your test center is well in advance of your test day so you do not get lost on the day of the test. On the night before the test, gather together the materials you will need the next day: - Your admission ticket - Two forms of identification (e.g., driver's license, student identification card, or current alien registration card) - Two No. 2 pencils with erasers - Directions to the test center - A watch (if you wish) but not one that makes noise, as it may disturb other test-takers On the day of the test, you should wake up early (after a good night's rest) and have breakfast. Dress comfortably, so that you are not distracted by being too hot or too cold while taking the test. Also, plan to arrive at the test center early. This will allow you to collect your thoughts and relax before the test, and will also spare you the stress of being late. If you arrive after the test begins, you will not be admitted to the test center and you will not receive a refund. During the Test When you arrive at the test center, try to find a seat where you feel most comfortable. Follow all the rules and instructions given by the test supervisor. If you do not, you risk being dismissed from the test and having your scores canceled. Once all the test materials are passed out, the test instructor will give you directions for filling out your answer sheet. Fill this sheet out carefully since this information will appear on your score report. After the Test When you have completed the SAT II: Chemistry Subject Test, you may hand in your test materials and leave. Then, go home and relax! When Will I Receive My Score Report and What Will It Look Like? You should receive your score report about five weeks after you take the test. This report will include your scores, percentile ranks, and interpretive information.

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Chemistry Tests Answer Key (for Use With)

O level chemistry multiple choice questions has 900 MCQs. GCSE chemistry quiz questions and answers, MCQs on IGCSE chemistry, electricity, acids, bases, chemical bonding, chemical formulas, chemical structure, chemical equations, physical chemistry, experimental chemistry MCQs with answers, chemicals, elements, compounds, mixtures, chemicals energy, purification methods, particles of matter, redox reactions, salts identification MCQs and quiz for SAT/ACT/GAT/GRE/CLEP/GED practice tests. GCSE, IGCSE chemistry multiple choice quiz questions and answers, chemistry exam revision and study guide with practice tests for SAT/ACT/GAT/GRE/CLEP/GED for online exam prep and interviews. Chemistry interview questions and answers to ask, to prepare and to study for jobs interviews and career MCQs with answer keys. Acids and bases quiz has 123 multiple choice questions. Chemical bonding and structure quiz has 75 multiple choice questions. Chemical formulae and equations quiz has 167 multiple choice questions with answers. Electricity and chemistry quiz has 108 multiple choice questions. Electricity and chemicals quiz has 10 multiple choice questions. Elements, compounds and mixtures quiz has 39 multiple choice questions. Energy from chemicals quiz has 41 multiple choice questions. Experimental chemistry quiz has 18 multiple choice questions. Methods of purification quiz has 84 multiple choice questions. Particles of matter quiz has 45 multiple choice questions. Redox reactions quiz has 42 multiple choice questions. Salts and identification of ions and gases quiz has 61 multiple choice questions. Speed of reaction quiz has 35 multiple choice questions. Structure of atom quiz has 52 multiple choice questions and answers. Chemistry interview questions and answers, MCQs on accounting acid rain, acidity needs water, acidity or alkalinity, acids properties and reactions, amphoteric oxides, applications of electrolysis, arrangement of particles in atom, atomic mass, atoms and elements, basic acidic neutral and amphoteric, catalysts and enzymes, change of state, chemical and ionic equations, chemical equations, chemical formulas, chemical reaction factor affecting, chemical reactions, chemical symbols, chemical to electrical energy, chemistry reactions, collection of gases, college chemistry, conductors and nonconductors, crystallization of microchips, decanting and centrifuging, dissolving, filtering and evaporating, distillation purification process, dry cells, electrical devices and circuit symbols, electrolyte and non-electrolyte, electrolytes and non-electrolytes, endothermic reactions, evaporation, exothermic reactions, fast and slow reactions, insoluble salts ionic precipitation, ionic and covalent substances, ionic compounds crystal lattices, ions and ionic bonds, isotopes number of neutrons, kinetic particle theory, kinetic theory, making and breaking bonds, mass, volume, time and temperature, measuring speed of reaction, method of purification, methods of purification sublimation, mineral acids general properties, mixtures and compounds, molar mass, molecules and compounds, molecules and covalent bonds, molecules and macromolecules, neutralization, states of matter, ordinary level chemistry, organic acid, organic solvents, oxidation and reduction, oxidation reduction reactions, paper chromatography, percent composition of elements, periodic table, PH scale acid and alkali, polarization, properties bases and reactions, proton and nucleon number, protons, neutrons and electrons, pure substances and mixtures, reactants, redox reaction oxidation, redox reactions, relative molecular mass, salts hydrogen of acids, save energy,

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Chemistry

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Level Chemistry MCQs: Multiple Choice Questions and Answers (Quiz & Tests with Answer Keys)

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Modern Chemistry

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Chemistry for the Health Sciences. Answer Key and Test Manual

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Let's Review Regents: Chemistry--Physical Setting Revised Edition

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and Theory MCQs" covers topics of atom facts, elements and atoms, number of nucleons, protons, electrons, and neutrons. The chapter "Benzene: Chemical Compound MCQs" covers topics of benzene, arenes reaction, phenol properties, and reactions of phenol. The chapter "Carbonyl Compounds MCQs" covers topics of carbonyl compounds, aldehydes and ketone testing, nucleophilic addition with HCN, preparation of aldehydes and ketone, reduction of aldehydes, and ketone.

McGraw-Hill's 500 MCAT General Chemistry Questions to Know by Test Day

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organic solvents, polarization, and transfer of electrons. The chapter "Chemical Formulae and Equations MCQs" covers topics of chemical formulas, chemical equations, atomic mass, ionic equations, chemical reactions, chemical symbols, mixtures and compounds, molar mass, percent composition of elements, reactants, relative molecular mass, valency and chemical formula, and valency table. The chapter "Electricity MCQs" covers topics of chemical to electrical energy, applications of electrolysis, reactions, conductors and non-conductors, dry cells, electrical devices, circuit symbols, electrolytes, non-electrolytes, organic solvents, polarization, and valence electrons. The chapter "Electricity and Chemicals MCQs" covers topics of chemical to electrical energy, dry cells, electrolyte, non-electrolyte, and polarization. The chapter "Elements, Compounds and Mixtures MCQs" covers topics of elements, compounds, mixtures, molecules, atoms, and symbols for elements.

NES Chemistry Secrets Study Guide

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TABLE OF CONTENTS

About Research & Education Association
Preface
About the Test Scoring
Contacting the AP Program
AP CHEMISTRY COURSE REVIEW

CHAPTER 1 - THE STRUCTURE OF MATTER

A. ATOMIC PROPERTIES

1. The Atomic Theory and Evidence for the Atomic Theory
2. Chemical and Physical Approaches to Atomic Weight Determination
3. Atomic Number and Mass Number, Isotopes, Mass Spectroscopy
4. Electron Energy Levels
5. The Periodic Table and Periodic Relationships: Symbols, Radii, Ionization Energy, Electron Affinity, Oxidation States

B. BONDING

1. Types of Bonds
2. Effects of Bonding Forces on States, Structures, and Properties of Matter
3. Polarity and Electronegativity
4. Geometry of Ions, Molecules, and Coordination Complexes
5. Molecular Models

C. NUCLEAR CHEMISTRY, NUCLEAR EQUATIONS, HALF-LIVES, RADIOACTIVITY

CHAPTER 2 - STATES OF MATTER

A. GASES

1. Ideal Gas Laws
2. Kinetic Molecular Theory

B. LIQUIDS AND SOLIDS

1. Kinetic-Molecular View of Liquids and Solids
2. Phase Diagram
3. Changes of State, Critical Phenomena
4. Structure of Crystals

C. SOLUTIONS

1. Types of Solutions
2. Factors Affecting Solubility
3. Ways of Expressing Concentrations
4. Colligative Properties
5. Interionic Attractions

CHAPTER 3 - REACTIONS

A. TYPES

1. Forming and Cleaving Covalent Bonds
2. Precipitation
3. Oxidation and Reduction

B. STOICHIOMETRY

1. Recognizing the Presence of Ionic and Molecular Species
2. Balancing Chemical Equations
3. Weight and Volume Relationships

C. EQUILIBRIUM

1. Dynamic Equilibrium Both Physical and Chemical
2. The Relationship Between K_p and K_c
3. Equilibrium Constants for Reactions in Solutions

D. KINETICS

1. Rate of Reaction
2. Reaction Order
3. Temperature Changes and Effect on Rate
4. Activation Energy
5. Mechanism of a Reaction

E. THERMODYNAMICS

1. State Functions
2. The First Law of Thermodynamics
3. The Second Law of Thermodynamics
4. Change in Free Energy

CHAPTER 4 - DESCRIPTIVE CHEMISTRY

1. Horizontal, Vertical, and Diagonal Relationships in the Periodic Table
2. Chemistry of the Main Groups and Transition Elements and Representatives of Each
3. Organic Chemistry
4. Structural Isomerism

PRACTICE EXAMS

AP CHEMISTRY EXAM I
AP CHEMISTRY EXAM II
AP CHEMISTRY EXAM III
AP CHEMISTRY EXAM IV
AP CHEMISTRY EXAM V
AP CHEMISTRY EXAM VI

FORMULAS AND TABLES EXCERPT

About Research & Education Association

Research & Education Association (REA) is an organization of educators, scientists, and engineers specializing in various academic fields. Founded in 1959 with the purpose of disseminating the most recently developed scientific information to groups in industry, government, high schools, and universities, REA has since become a successful and highly respected publisher of study aids, test preps, handbooks, and reference works. REA's Test Preparation series includes study guides for all academic levels in almost all disciplines. Research & Education Association publishes test preps for students who have not yet completed high school, as well as high school students preparing to enter college. Students from countries around the world seeking to attend college in the United States will find the assistance they need in REA's publications. For college students seeking advanced degrees, REA publishes test preps for many major graduate school admission examinations in a wide variety of disciplines, including engineering, law, and medicine. Students at every level, in every field, with every ambition can find what they are looking for among REA's

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ABOUT THE TEST The Advanced Placement Chemistry Examination is offered each May at participating schools and multi-school centers throughout the world. The Advanced Placement Program is designed to allow high school students to pursue college-level studies while attending high school. The participating colleges, in turn, grant credit and/or advanced placement to students who do well on the examinations. The Advanced Placement Chemistry course is designed to be the equivalent of a college introductory chemistry course, often taken by chemistry majors in their first year of college. Since the test covers a broad range of topics, no student is expected to answer all of the questions correctly. The exam is divided into two sections: 1) Multiple-choice: Composed of 75 multiple-choice questions designed to test your ability to recall and understand a broad range of chemical concepts and calculations. This section constitutes 45% of the final grade and you are allowed 90 minutes for this portion of the exam. Calculators are not permitted for this section of the exam. 2) Free-response section: Composed of several comprehensive problems and essay topics. This section constitutes 55% of the final grade and the student is allowed 90 minutes for this portion of the exam. You may choose from the questions provided. These problems and essays are designed to test your ability to think clearly and to present ideas in a logical, coherent fashion. You can bring an electronic hand-held calculator for use on the 40-minute free-response section. Essay and chemical-reaction questions comprise the last 50 minutes of the test, during which calculators are not permitted. A final note about calculators: Most hand-held models are allowed in the test center; the only notable exceptions are those with typewriter-style (QWERTY) keypads. If you are unsure if your calculator is permitted, check with your teacher or Educational Testing Service.

SCORING The multiple-choice section of the exam is scored by crediting each correct answer with one point, and deducting only partial credit (one-fourth of a point) for each incorrect answer. Omitted questions receive neither a credit nor a deduction. The essay section is scored by a group of more than 1,000 college and high school educators familiar with the AP Program. These graders evaluate the accuracy and coherence of the essays accordingly. The grades given for the essays are combined with the results of the multiple-choice section, and the total raw score is then converted to the program's five-point scale: 5 - Extremely well qualified 4 - Well qualified 3 - Qualified 2 - Possibly qualified

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- Tons of sample problems and drills with detailed explanations

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GACE Chemistry Study Guide

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