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Intro
What are coefficients
What are molar ratios
Mole mole conversion
Mass mass practice

Titration: Practical and Calculation (NaOH and HCl) - Titration: Practical and Calculation (NaOH and HCl) by FranklyChemistry 948,637 views 11 years ago 8 minutes, 56 seconds - This video, with the help of flash animations, shows and explains how you can determine the concentration of a sodium hydroxide ...

touch the surface of the liquid in the flask
read the bottom of the meniscus
discard the chemicals in the flask
wash off the half drop from the bottom of the burette
rinse it well with deionized water
obtain an accurate titer
get the number of moles in the flask
get the concentration of the chemical in the flask
work out the number of moles of sodium hydroxide
concentration in the flask cover

Solving Acid-Base Titration Problems - Solving Acid-Base Titration Problems by Danielle Paige 319,639 views 10 years ago 13 minutes, 24 seconds - A step-by-step tutorial on solving acid-base titration math problems. Uses the double mole map method focusing on 4 steps: 1.

Mole Ratio
Mole Map
Balancing Chemical Equations
Sample Problem
Concentration Problem

Limiting Reactant Practice Problem - Limiting Reactant Practice Problem by Tyler DeWitt 1,201,208 views 8 years ago 10 minutes, 47 seconds - We'll practice limiting reactant and excess reactant by working through a problem. These are often also called limiting reagent and ...

starting with a maximum amount of magnesium
figure out the greatest amount of magnesium oxide
start with a maximum amount of the limiting reactant
start with the total reactant

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Statistics in Analytical Chemistry (Part 2) by Mr. ANRA 7,057 views 2 years ago 26 minutes - Exercise problems: 0:00 - 1. The results of an analysis are 36.97 grams, compared with the accepted value of 37.06 grams.

1. The results of an analysis are 36.97 grams, compared with the accepted value of 37.06 grams.

What is the relative error in parts per thousand?

2. Calculate the mean and the standard deviation for the set of measured values: 15.67, 15.69, 16.03

3. Which in the following sets of data is more precise

4. The carbohydrate content of a glycoprotein is determined to be 12.6, 11.9, 13.9, 12.7, and 12.5 of carbohydrate per 100g of protein. Find the confidence interval of the carbohydrate content from this analysis.

AQA A-Level Chemistry - Amount of Substance Pt. 1 (moles, concentrations and masses) - AQA

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What a Mole Is

Converting Volumes

Question Number One

Titration

Calculating the Concentration

Calculate the Moles of Sodium Hydroxide

Calculate Concentration in Part C

Work Out the Moles of Hydrochloric Acid

A LEVEL CHEMISTRY EXAM QUESTION WALKTHROUGH - AS ANALYTICAL TECHNIQUES 1

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1 by MaChemGuy 6,910 views 1 year ago 4 minutes, 33 seconds - Link to question <https://drive.google.com/file/d/1sC93KZ9SPuRD2RqvCYRyadcW5au4VeOt/view?usp=drivesdk>.

Gathering the Evidence

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Introduction

Balancing Equations

Relative Formula Mass

Percentage by Mass

Moles

Using the Moles Equation

Balancing Equations with moles

Reacting Mass Calculations

Limiting Factor and Excess

Using Limiting Factor

Concentration

Grams per decimetre cubed

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looking for the concentration of the original hcl solution

find the moles of sodium hydroxide

start with the molarity of sodium hydroxide

move the decimal point three units to left

find the concentration

keep in mind the moles of the acid

plug in the information of the base
write point 2.9 moles of nitric acid per liter
get rid of unit moles of nitric acid
convert liters in to milliliters
moles of naoh
multiply that by the volume of the naoh solution
convert the moles of khp into grams using the molar mass
find a concentration of koh

CH403 5 Quality Assurance and Calibration Methods - CH403 5 Quality Assurance and Calibration Methods by Ratliff Chemistry 21,358 views 8 years ago 18 minutes - A brief introduction to quality assurance and why it might matter to chemists. This is from Chapter 5 in **Quantitative Chemical**, ...

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